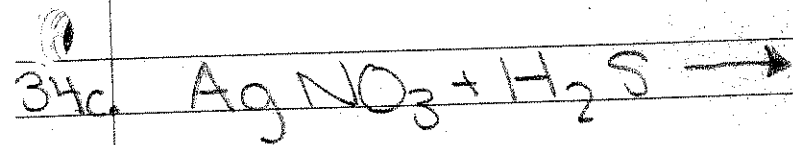
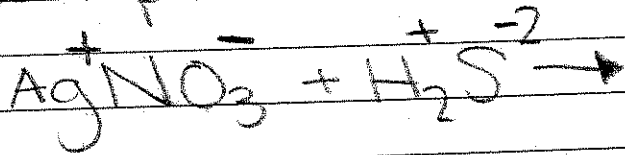


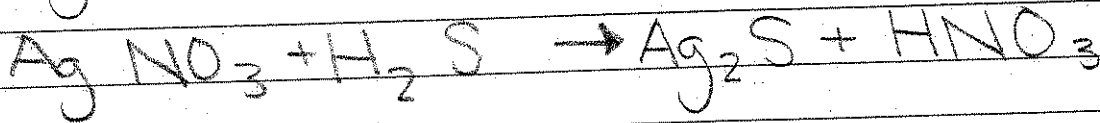
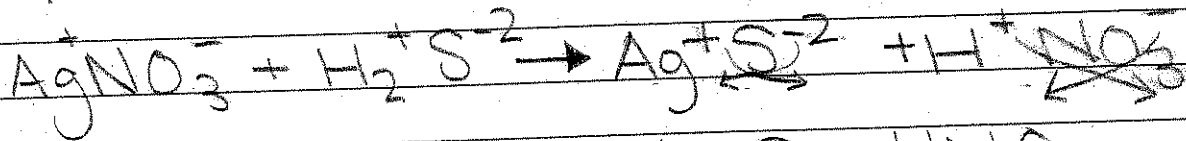
# Net ionic equation & precipitate (344)



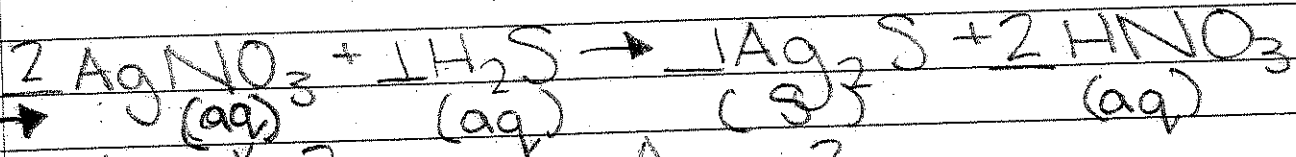
① Find charges of each part of the compound



② Switch partners  
\* ignore all subscripts unless part of a polyatomic ion \*



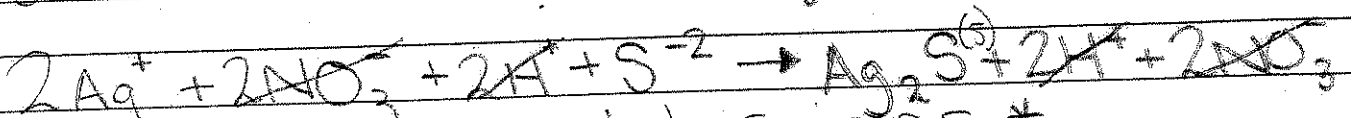
③ balance equation



Ag	x 2	Ag	2
N	x 2	N	x 2
O	3 6	O	3 6
H	2	H	x 2
S	1	S	1

④ find which compounds are (aq) or (s)  
(use solubility chart)

⑤ write net ionic equation



⑥ cancel out spectator ions \*

